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## Lewis dot structure practice hard

Spreadsheets lewis dot structures from the Department of Chemistry and Physics at Arkansas State University Draw Lewis Dot Structures for each of the following, enter electron arrangement (E.A.) and molecular geometry (M.G.): PF5 CS2 BrO3- NH4+ SiCl4 BrF5 BF3 SCI6 PH3 NF3 SO42- CO32- SiCl4 ClO3- CH2O NO3- O3 CCl4 AlH3 SO2 SO3 CO CHCl3 BRF3 H2S I3- H3O+ responds PROBLEM [\\(\PageIndex{1}\\)](#) Write Lewis structures for: (note, that none of the solutions use the extended octet rule or official charges) H2 HBr PCI3 SF2 H2CCH2 HNNH H2CNH NO- N2 CO CN- Answer answer b Reply c Reply e Reply e Reply g Response h Answer j Answer k PROBLEM [\\(\PageIndex{2}\\)](#) Type Lewis structures as follows: (note, none of the solutions use extended octet rule or formal costs) O2 H2CO CINO SiCl4 H3O+  $\text{NH}_4^+$   $\text{BF}_4^-$  HCCH CLCN  $\text{C}_2^{2+}$  Reply b Reply c Reply d Reply g Reply g Reply h Reply j PROBLEM [\\(\PageIndex{3}\\)](#) Type Lewis Structures : (note, none of the solutions use an extended octet rule or formal costs) a.  $\text{PO}_4^{3-}$  c.  $\text{SO}_3^{2-}$  d. HONO Respond to the answer b Reply c Click here to see a video of the solution PROBLEM [\\(\PageIndex{4}\\)](#) Methanol , H3COH, used as fuel in some racing cars. Ethanol, C2H5OH, is widely used as motor fuel in Brazil. Both methanol and ethanol emit carbon dioxide and H2O when burned. Write chemical equations for these combustion reactions using Lewis' structures instead of chemical formulas. Answer problem [\\(\PageIndex{5}\\)](#) Many planets in our solar system contain organic chemicals such as methane (CH4) and residues of etane (C2H4), etane (C2H6), propyne (H3CCCH) and diasetylene (HCCCCH). Write Lewis' structures for each molecule. (diasetylene can be a little tricky!) Answer Click here to see a video of the solution PROBLEM [\\(\PageIndex{6}\\)](#) Carbon tetrachloride was previously used from electrical fire extinguishers. It is no longer used for this purpose due to the formation of the toxic gas phosgene Cl2CO. Write Lewis' structures on carbon tetrachloride and phosgene. Answer PROBLEM [\\(\PageIndex{7}\\)](#) The arrangement of atoms in several biologically important molecules is here. Complete the Lewis structures of these molecules by adding several bonds and lonely pairs. Don't add atoms. a. amino acid serine: b. urea: c. pyruvic acid: d. uracil: e. carbonic acid: Reply b Answer b Answer d Answer e PROBLEM [\\(\PageIndex{8}\\)](#) How are individual, double and triple bonds similar? How do they differ? Answer Each bond involves sharing electrons between atoms. Two electrons are divided together into a bond; four electrons are distributed in a double sensation; and six electrons are divided into triple bandages. Participants are the answers above wrong? Just tell us. Enter the name of the electronic arrangement and the name of the molecular geometry of each species #3. Check your answer If you see this post, it means we have problems uploading external resources to our website. If you are behind a web filter, make sure that domains \*.kastatic.org \*.kasandbox.org are blocked. If you see this post, it means that we have problems downloading external resources on our website. If you are behind a web filter, make sure that domains \*.kastatic.org \*.kasandbox.org are blocked. prevented.

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