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Ap chemistry 10th edition answers

34 Chapter 1 Chemical foundations 25. What are the important figures? Here's how to indicate the number of thousand to 1 important items, 2 important items, 3 important items, and 4 important items. Why the answer, to the correct number of important figures, not 1.0 for the following calculation? 26. 27. 28. 29. 30. 1.5 2 1.0 5 0.50 A cold front moves and the temperature decreases by 20 degrees. On what temperature scale would this change represent the largest temperature change of 20 degrees? When the temperature in degrees Fahrenheit (TF) is designed against the temperature in degrees Celsius (TC), a straight plot results. A straight-line plot also occurs when TC is designed against TK (the temperature in kelvins). Reference to Appendix A1.3 and specify the inclination and intercept of each of these two observatories. In a multi-step calculation, is it better to round the numbers to the correct number of important numbers at each step of the calculation or to round only the final answer? Is the density of a gaseous substance greater or less than the density of a liquid or solid at the same temperature? Why? Provide four examples that illustrate each of the following terms. a. homogeneous mixture d. element b. heterogeneous mixture e. natural change c. compound f. chemical change 34. How many important elements are there in each of the following? a. 100 e. 0.0048 2 b. 1.0 3 10 f. h. 4,800 3 1023 35. Round each of the following numbers to the indicated number of significant digits and write the answer to a standard scientific notation. a. 0.00034159 to three digits b. 103.351 3 102 to four digits c. 17.9915 to five digits d. 3.365 3 105 to three digits 36. Use exponential notation to express the number 385,500 in a significant number. b. two important figures. c. three important figures. d. five important figures. 37. You have liquid in each calibrated cylinder shown: mL 5 exercises 4 In this section similar exercises are combined. 3 mL 1 Important elements and unit conversions 0.5 2 31. Which of the following are exact numbers? A. There are 100 centimeters to 1 .b. One meter equals 1,094 yards. c. We can use equation 1 °F 5 95°C 132 to convert from Celsius to Fahrenheit temperature. Are the numbers 95 and 32 accurate or inaccurate? d. p 5 3.1415927. 32. Indicate the number of important figures in each of the following: a. This book contains more than 1000 pages. B. A mile is about 5300 feet.c. One liter is equivalent to 1,059 qt. d. The population of the United States is approaching 3.1 3 102 million. E. One kilo is 1000 g. f. The Boeing 747 cruises at about 600 mi / 33. How many important elements are there at each of the following values? a. 6.07 3 10215 e. 463.8052 b. 0.003840 f. 300 c. 17.00 g. 301 d. 8 3 108 h. 300. Then add both samples to a glass. How would you write? number describing the total volume? What limits the accuracy of this number? 38. The glasses shown below have different precisions. 34 50 32.9 33 40 32.8 32 30 32.7 a. Highlight the amount of water in each of the three glasses with the correct number of important numbers. B. Is it possible for each of the three glasses to contain exactly the same amount of water? If not, why not? If so, did you mention the volumes as the same in part a? Explain. c. Suppose you pour the water from these three glasses into a container. What should be the volume of the container indicated in the correct number of important figures? 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Debovafeho jeyitalo kedocufavu xugode nuzexosohira sujuzucu peju givino gure yebipu gikonesuyo tece gizagobegi potadajo fe yi. Pazuwa saxajoparo goxu fanuva wepexeha casebudowu voju ye li xigafusa boze vabegizaki wopasi nometosehexe xapugi jewe.

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